Preliminary Phytochemical Analysis and TLC Fingerprinting of Desmostachya bipinnata

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Abstract

The nature has provided an entire warehouse of remedies to cure ailments of mankind. About 80 per cent of the world's population depends partially or wholly on traditional medicine for its primary health care needs. Herbal medicines as the key remedy in traditional medical system have been used in medical practice for past era and have made a great involvement to maintain human health. Thus Pharmacognostical study appears to be of enormous value in identification of commercial samples of the market to find their authenticity and establishing identity of adulterant or substituent. In addition lack of standardization, dishonest commercial practice of adulterating and substituting the genuine herbal drugs are posing great obstacle in popularizing the time-tested herbal-based traditional medicines. The present study was focused on preliminary pharmacognostic and phytochemical investigations for determining and establishing the identity, purity and quality of the plants. The plant was subjected to determination of various physicochemical parameters including ash values (total ash, water soluble ash) and extractive values (alcohol soluble extractive, water soluble extractive). The powdered crude drug was extracted successively with various solvents with increasing polarity and further the extracts were subjected to phytochemical screening for the identification of various phytoconstituents. The study of heavy metal analysis, total bacterial count & test for Aflatoxins studies was also carried out as WHO guidelines. More over the fingerprint profile of Desmostachya bipinnata was established using thin layer chromatography(TLC) methods. The results of the TLC profiling of the extract confirm about the presence of various phytochemicals.

Keywords: of Desmostachya bipinnata, Microscopy, Phytochemical investigations & TLC profiling.

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INTRODUCTION

An herbal drug constitutes a most important part in all traditional systems of medicine. There are approximately 1250 Indian medicinal plants which are used in formulating therapeutic preparations according to Ayurvedic and other traditional systems of medicine. Plants endow with varieties of resources that contribute to the fundamental need of food, clothing and shelter[1]. Desmostachya bipinnata (Linn.) belongs to the family, Poaceae (Graminaceae). It is well-known as Sacrificial Grass or Saved Gram in English. The plant is known for holy purpose as well as medicinally uses. It is famous as Tharuppai and its kudineer is prescribed for any type of disorder, fevers, itching and diuretic problems in Siddha literatures. It is valuable for curing urinary tract diseases and excessive vaginal discharges [2]. Many secondary metabolites such as scopoletine, umbelliferone, sugars, amino acids, carbohydrates are reported in the plant. Desmostachya bipinnata is traditionally known as Kusha grasses that are form vital ingredient in various Vedic sacrifices (Yagnas) and rituals. They are found along river beds and plains throughout India and are well known for their medicinal properties as supported by traditional Ayurvedic scriptures and are a vital component in traditional medicinal formulations such as Tripanchamool, Kusadya-ghrita and Kusa blecha, etc. to treat many disorders such as dysentery, diuresis, jaundice, skin infections etc[3].
MATERIALS AND METHODS

Collection

The sample of Desmostachya bipinnata was collected from Haridwar, State – Utrakhand. The sample of Desmostachya bipinnata was collected by scholar under the guidance of Supervisor after identifying the source of plant as per standard description.

Date of Collection

Desmostachya bipinnata-5/12/2015 (Haridwar)

Authentication

The plant was authentication was done Botanical Survey of India (BSI), Dehradun.

Microscopic study

Microscopic study of crude drugs is another aid of Pharmacognosy which can be helpful in the process of standardization of medicinal plants. This study can be helpful in identifying genuine drug by their known histological characters through Transverse section (T.S.) or Longitudinal Section (L.S.) or Radial Longitudinal Section (R.L.S.) or Tangential Longitudinal Section (T.L.S.) and Powder microscopy which can help in evaluation of different constituents by using different staining reagents. Specimens were soaked in water or other solvents depending upon the hardness of the sample and transverse sections were taken using sharp razor blades. Few microscopic sections were cut by Microtome sectioning. Numerous temporary and permanent mounts of the microscopical sections of the specimen were made and examined microscopically. Different staining reagents were applied on transverse sections so as to differentiate between different cell wall components.

Preliminary physiochemical screening of plant extracts[4-8]

Determination of solvent extractive values

Determination of water soluble extractive value: 5 g of the air-dried drug, coarsely powdered were macerated with 100 ml of water in closed flask for 24 hours, shaking frequently during the first 6 hours and allow standing for 18 hours. It was filtered rapidly taking precaution against loss of water, then the filtrate was evaporated 25 ml of the filtrate to dryness in a tared flat-bottomed shallow dish, and dried at 105°C then weighed. The percentage of water-soluble extractive with reference to the air dried was calculated.

Determination of alcohol soluble extractive value: 5 gm of the air dried and coarsely powdered drug was macerated with 100 ml of ethanol of the specific strength in a closed flask for 24 hours, shaking frequently during the first 6 hours and allow standing for 18 hours. There after filter rapidly taking precaution against loss of ethanol. Evaporate 25 ml of the filtrate to dryness in a tared flat bottomed shallow dish, dry at 105°C and weigh. The percentage of ethanol soluble extractive with reference to the air dried drug has to be calculated.

Determination of moisture content: Moisture is an inevitable component of crude drugs, which must be eliminated as far as practicable. Method of determination of moisture content include the loss on drying, the test for loss on drying determines both water and volatile matter in the crude drug. It can be carried out either by heating at 100°C-105°C or in a dessicator over phosphorous pentoxide under atmospheric or reduced pressure at room temperature for specific period of time.

Ash value: Ash value is helpful in determining the quality and purity of a crude drug, especially in the powdered form. On incineration, crude drugs normally leave an ash usually consisting of carbonates, phosphates and silicates of sodium, potassium, calcium and magnesium. The total ash of a crude drug reflects the care taken in its preparation. A higher limit of acid-insoluble ash is imposed, especially in case where silica may be present or when the calcium oxalate content of the drug is very high.

Total ash value: Weighed accurately about 2 to 3 g of the powdered drug in a tared silica crucible. Incinerated at a temperature not exceeding 450 °C for 4 hr, until free from carbon, cooled and weighed. The percentage of ash with reference to air-dried was calculated following formula.

% Total ash value = \[ \frac{\text{Wt. of total ash}}{\text{Wt. of crude drugs}} \times 100 \]
Water soluble ash value: Boiled the ash with 25 ml of water. Filtered and collected the insoluble matter on an ash less filter paper, washed with hot water and ignited in a tared crucible at a temperature not exceeding 450°C for 4 hr. Cooled in a desiccator and weighed. Subtract the weight of insoluble matter from the total weight of ash. The difference in weight represented weight of water soluble ash. Calculated the percentage of water soluble ash with reference to the air-dried drug by using the following formula.

\[
\% \text{ Water soluble ash value} = \frac{\text{Wt. of total ash} - \text{Wt. of water insoluble ash}}{\text{Wt. of crude drug taken}} \times 100
\]

Acid insoluble ash value: Boiled the ash for 5 min with 25 ml of 2 M HCl. Filtered and collected the insoluble matter on an ash less filter paper, washed with hot water and ignited in a tared crucible at a temperature not exceeding 450°C for 4 hr. Cooled in a desiccator and weighed. Calculated the percentage of acid insoluble ash with reference to the air-dried drug was calculated by using following formula,

\[
\% \text{ Acid insoluble ash value} = \frac{\text{Wt. of acid insoluble ash}}{\text{Wt. of crude drug taken}} \times 100
\]

Preliminary qualitative test

The various extract of *Desmostachya bipinnata* was subjected to preliminary qualitative phytochemical investigation. The various tests and reagent used are given below.

**Alkaloids**

Preparation of test solution: The test solution was prepared by dissolving extracts in the dilute hydrochloric acid.

Mayer’s test: The acidic test solution with Mayer’s reagent (Potassium Mercuric iodide) gave cream colored precipitate.

Hager’s test: The acidic test solution with Hager’s reagent (Saturated picric acid solution) gave yellow precipitate.

Dragendorff’s test: The acidic solution with Dragendorff’s reagent (Potassium bismuth iodide) showed reddish brown precipitate.

Wagner’s test: The acidic test solution treated with Wagner’s reagent (Iodine in potassium iodide) gave brown precipitate.

Tannic acid test: The acidic test solution treated with Tannic acid gave buff colour precipitate.

Picrolonic acid test: Alkaloids gave yellow colour precipitate with picrolonic acid.

**Amino acid**

Millon’s test: To the test solution add about 2 ml of millon’s reagent white precipitate indicates presence of amino acid.

Ninhydrine test: To the test solution add Ninhydrine solution, boil, violet colour indicates presence of amino acid.

**Carbohydrates**

Preparation of test solution: The test solution was prepared by dissolving the test extracts with water. Then it was hydrolyzed with 1 volume of 1 N-HCl and subjected to following chemical test.

Molisch’s test: Test solution with few drops of Molisch’s reagent and 2 ml of conc. H₃SO₄ added slowly from the sides of the test tubes. It showed a purple ring at the junction of two liquids.

Barfoed’s test: 1 ml of test solution is heated with 1 ml of Barfoed’s reagent on water bath, if red cupric oxide is formed, monosaccharide is present. Disaccharides on prolong heating (about 10 min.) may also cause reduction, owing to partial hydrolysis to monosaccharide.

Benedict’s test: Test solution treated with Benedict’s reagent and after boiling on water bath, it showed reddish brown precipitate.

Fehling’s test: The test solution when heated with equal volume of Fehling’s A and B solution, gave orange red precipitate, indicating the presence of reducing sugars

Flavonoids: The flavonoids are all structurally derived from the parent substance called flavones. The flavonoids occur in the free from as well as bound to sugars as glycosides. For this reason, when analyzing flavonoids it is usually better to examine the flavonoids in hydrolyzed plant extracts.
Preparation of test solution: To a small amount of extract added equal volume of 2 M HCL and heated in a test tube for 30 to 40 min at 100°C. The cooled extract was filtered, and extracted with ethyl acetate. The ethyl acetate was concentrated to dryness, and used to test for flavonoids.

Shinoda test: Test solution with few fragments of magnesium ribbon and conc. HCL showed pink to magenta red colour. To a small quantity of test solution when lead acetate solution was added, it formed yellow colored precipitate.

Alkaline reagent test: Test solution when treated with sodium hydroxide solution showed increase in the intensity of yellow colour, which becomes colorless on addition of few drops of dilute acid.

**Glycosides**

Preparation of test solution: The test solution was prepared by dissolving extract in the alcohol or hydro-alcoholic solution.

**Test for Cardiac glycosides**

Kedde’ test: Add one drop of 90% alcohol and 2 drops of 2 % 3, 5- dinitro benzoic acid in 90% alcohol. Make alkaline with 20 % sodium hydroxide solution, purple colour is produced. The colour reaction with 3, 5- dinitro benzoic acid depends on the presence of α, β-unsaturated lactones in the aglycone.

Baljet’s test: The test solution treated with sodium picrate gave yellow to orange colour.

Raymond’s test: Test solution treated with hot methanolic alkali, violet colour is produced. Bromine water test: Test solution dissolve in bromine water give yellow precipitate.

Keller-killani test for digitoxose: The test solution treated with few drops of FecI3 solution and mixed, then H2SO4 containing FecI3 solution was added, it formed two layers. Lower layer reddish brown, upper layer turns bluish green.

Legal’s test: Test solution when treated with pyridine (made alkaline by adding sodium nitroprusside solution) gave pink to red colour.

**Test for anthraquinone glycosides**

Borntrager’s test: Boiled powdered drug with 5 ml of 10 % sulphuric acid for five minutes. Filtered while hot, cooled the filtrate shaken gently with equal volume of benzene. Benzene layer was separated and then treated with half of its volume solution ammonia (10%). Allowed to separate it. The ammonical layer acquired rose pink colour due to presence of anthraquinones.

**Proteins**

Preparation of test solution: The test solution was prepared by dissolving the extract in water.

Millon’s test: Test solution was treated with millon’s reagent and heated on a water bath. The proteins were stained red

Biuret test: Test solution was treated with 40% sodium hydroxide and dilute copper sulphate solution gave blue colour.

Xanthoproteic test: Test solution was treated with conc. HNO3 and boiled which gave yellow precipitate.

Modified Borntrager’s test: C-glycosides of anthraquinones require more drastic conditions for hydrolysis. Hydrolysis of the drug was carried out with 5 ml of dilute of HCL and 5 ml of 5 % solution of FeCl3.

For hydrolyzed extract procedure was carried out as described under Borntrager’s test.

**Test for steroids**

Preparation of test extract solution: The extract was refluxed separately with alcoholic solution of potassium hydroxide till complete saponification. The saponified extract was diluted with water and unsaponifiable matter was extracted with diethyl ether.

The ethereal extract was evaporated and the residue (saponifiable matter) was subjected to the following test by dissolving the residue in the chloroform.

Salkowski test: To the test extract solution add few drops of conc. H2SO4 shaken and allowed to stand, lower layer turned red indicating the presence of steroids.

Libermann - Burchard test: The test solution treated with few drops of acetic anhydride and mixed, when conc. H2SO4 was added from the sides of the test tubes, it showed a brown ring at the junction of the two layers and the upper layers turned green. Added few drops of concentrated H2SO4. Blue colour appeared.

**Sulphur test:** Sulphur test when added in to the test solution, it sank it.

**Tannins and phenol compound**

To 2-3 ml of alcoholic or aqueous extract, added few drops of following reagents.

5% FeCl3 solution: Deep blue- black colour.

Lead acetate solution: White precipitate.

Bromine water: Discoloration of bromine water

Acetic acid solution: Red colour solution.
Dilute iodine solution: Transient red colour

One drop of NH₂OH, excess 10% AGNO₃ solution. Heated for 20 min in boiling water bath. White precipitate was observed, then dark silver mirror deposited on wall of test tube.

Triterpenoids

Preparation of test extract solution: The test extract solution was prepared by dissolving extract in the chloroform.

Salkowski test: Few drops of concentrated sulphuric acid were added to the test solution, shaken and on standing lower layer turned golden yellow.

Microbiological determination tests [9-10]

Total viable aerobic count (TVC): Detection of the anti-bacterial activity of the test drug, the total viable aerobic count (TVC) of the test drug was carried out, as specified in the test procedure, using plate count. Pre-treatment of the test drug Depending on the nature of the herbal drug sample used, it was dissolved using a suitable method and any antimicrobial property present in the sample was eliminated by dilution or neutralization. Buffered Sodium Chloride-Peptone Solution, pH 7.0 was used to dilute the test sample.

Plate count for bacteria and fungi

For bacteria: 1 ml of the pretreated test sample was added to about 15 ml of the liquefied casein-soybean digest agar in a petridish of 90 mm diameter at a temperature not exceeding 45°C. Alternatively the test sample was spread on the surface of the solidified medium. Two dishes were prepared with the same dilution, they were inverted and incubated at 30-35°C for 48-72 h, unless a more reliable count was obtained in a short period of time. The number of colonies so formed was counted and the results were calculated using the plates with the largest number of colonies, up to a maximum of 300.

For fungi: 1 ml of the pretreated test sample was added to about 15 ml of the liquefied Sabereds glucose agar with antibiotics in a petridish of 90 mm diameter at a temperature not exceeding 45°C. Alternatively the test sample was spread on the surface of the solidified medium. Two dishes were prepared with the same dilution; they were inverted and incubated at 20 – 25°C for 5 days, unless a more reliable count was obtained in a short period of time. The number of colonies so formed was counted and the results were calculated using the plates with not more than 100 colonies.

Aflatoxins

The test for determination of the aflatoxins was carried out using LCMS-Ms.

TLC analysis for different Phytochemicals

T.L.C. plate coated with 0.25 mm layer of silica gel GF₂₅₄ with fluorescent indicator, (Mercks) were used. (Each plate dimension is 10 cm long and 2 cm width).

Activation of pre-coated Silica gel G60F₂₅₄

Dry in hot oven at 105°C for one to two hour.

Solvent system: Toluene, Ethyl Acetate (7:3)

Sample: Alcoholic Extract of test sample

Visualization: p-Anisaldehyde – sulfuric acid Reagent.

RESULT AND DISCUSSION

In Vedic texts Trinpanchmoola drugs like Kush is called ideal grass in Hindu religion, and described as Asan for Indra. The roots are cooling, diuretic, astrigent and galactagogue and are useful inn asthma, jaundice, stimulant, acrid and aphrodisiac and useful in dysentery, menorrhagia, jaundice, asthma and skin erosion. The collected drug were studied organoleptically with naked eye & magnifying lens, with the help of different Organoleptic features i.e. Colour, Odour, Taste, and Appearance were recorded(table 1). T.S of root showing sclerenchymatous cells,endodermis, metaxylem and phloem. The powder microscopic characteristics study revealed the presence of tannins, calcium sulphate ,cutin & lignins (table 2). Various physiochemical parameters were studied and result was tabulated in the table 3. The extractive value was found to be 11.21% w/w % (Alcohol Extractive Value) and 16.89% w/w (Aqueous Extractive Value).The phytochemical screening(table 4) revealed the presence of the various phytochemicals in order Aqueous extract > Alcoholic extract > Pet.ether extract. As per WHO guideline of quality control of Herbal plants, the heavy metals analysis, total bacterial count and aflatoxins detection studies were carried out. The result was tabulated in the table 5-7. In Chromatographic study, TLC (table 8) study showed the presence of phenol, sugars, steroids and terpenes nature of chemical constituents in all genuine samples and Rf value of Desmostachya bipinnata were 0.15,0.78,0.81,0.87.

CONCLUSION

The present study deals with macroscopic, powder microscopic, phytochemical analysis and TLC fingerprinting. These studied will make available referentially information for correct identification and for standardization. This study also gives support to in the assessing of adulteration and substitution in market sample of the plant.
### Table-1: Organoleptic Features

<table>
<thead>
<tr>
<th>Characters</th>
<th>Desmostachya bipinnata</th>
</tr>
</thead>
<tbody>
<tr>
<td>Appearance</td>
<td>Tufted grass with a thick scaly root stock</td>
</tr>
<tr>
<td>Colour</td>
<td>Pale brown</td>
</tr>
<tr>
<td>Odour</td>
<td>Odourless</td>
</tr>
<tr>
<td>Taste</td>
<td>Slight bitter</td>
</tr>
</tbody>
</table>

### Table-2: Powder Microscopy

![Powder Microscopy Image]

### Table-3: Physiochemical study of Parameters

<table>
<thead>
<tr>
<th>S.No</th>
<th>Tests</th>
<th>Desmostachya bipinnata</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Moisture content</td>
<td>8.70% w/w</td>
</tr>
<tr>
<td>2</td>
<td>pH</td>
<td>7.97</td>
</tr>
<tr>
<td>3</td>
<td>Alcohol Extractive Value</td>
<td>11.21% w/w</td>
</tr>
<tr>
<td>4</td>
<td>Aqueous Extractive Value</td>
<td>16.89% w/w</td>
</tr>
<tr>
<td>5</td>
<td>Pet.ether Extractive value</td>
<td>3.74% w/w</td>
</tr>
<tr>
<td>6</td>
<td>Foreign matter</td>
<td>1.14% w/w</td>
</tr>
<tr>
<td>7</td>
<td>Total Ash</td>
<td>6.82% w/w</td>
</tr>
<tr>
<td>8</td>
<td>Acid Insoluble Ash</td>
<td>2.45% w/w</td>
</tr>
<tr>
<td>9</td>
<td>Water Soluble Ash</td>
<td>3.13%w/w</td>
</tr>
</tbody>
</table>
### Table-4: Results of preliminary phytochemical analysis

<table>
<thead>
<tr>
<th>Sr. No.</th>
<th>Name of the Test</th>
<th>Observation</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>Aqueous Extract</td>
</tr>
<tr>
<td>1</td>
<td>Tests for sterols</td>
<td>+</td>
</tr>
<tr>
<td></td>
<td>1. Salkowski’s Test</td>
<td></td>
</tr>
<tr>
<td></td>
<td>2. Libermann Burchard’s Test</td>
<td>+</td>
</tr>
<tr>
<td>2</td>
<td>Test for glycosides</td>
<td>+</td>
</tr>
<tr>
<td></td>
<td>1. Baljet’s Test</td>
<td></td>
</tr>
<tr>
<td></td>
<td>2. Brontrager Test</td>
<td>-</td>
</tr>
<tr>
<td>3</td>
<td>Tests for saponins</td>
<td>+</td>
</tr>
<tr>
<td></td>
<td>1. Foam Test</td>
<td></td>
</tr>
<tr>
<td>4</td>
<td>Test for carbohydrates</td>
<td>+</td>
</tr>
<tr>
<td></td>
<td>1. Molish’s Test</td>
<td></td>
</tr>
<tr>
<td></td>
<td>2. Barfoed’s Test</td>
<td>+</td>
</tr>
<tr>
<td></td>
<td>3. Benedict’s Test</td>
<td>+</td>
</tr>
<tr>
<td>5</td>
<td>Tests for alkaloids</td>
<td>+</td>
</tr>
<tr>
<td></td>
<td>1. Mayer’s Test</td>
<td></td>
</tr>
<tr>
<td></td>
<td>2. Wagner’s Test</td>
<td>-</td>
</tr>
<tr>
<td></td>
<td>3. Dragendorff’s Test</td>
<td>+</td>
</tr>
<tr>
<td>6</td>
<td>Tests for flavonoids</td>
<td>+</td>
</tr>
<tr>
<td></td>
<td>1. Ferric chloride Test</td>
<td></td>
</tr>
<tr>
<td></td>
<td>2. Shinoda Test</td>
<td>+</td>
</tr>
<tr>
<td></td>
<td>3. Alkaline Reagent Test</td>
<td>+</td>
</tr>
<tr>
<td></td>
<td>4. Lead Acetate Test</td>
<td>+</td>
</tr>
<tr>
<td>7</td>
<td>Tests for tannins</td>
<td>+</td>
</tr>
<tr>
<td></td>
<td>1. Ferric chloride Test</td>
<td></td>
</tr>
<tr>
<td></td>
<td>2. Gelatin Test</td>
<td>+</td>
</tr>
<tr>
<td>8</td>
<td>Test for amino acid and protein</td>
<td>+</td>
</tr>
<tr>
<td></td>
<td>1. Biurete test</td>
<td></td>
</tr>
</tbody>
</table>

### Table-5: Heavy metal analysis

<table>
<thead>
<tr>
<th>S.No.</th>
<th>Heavy metal</th>
<th>Desmostachya bipinnata (S1)</th>
<th>Permissible limit</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Lead</td>
<td>Not Detected</td>
<td>10ppm</td>
</tr>
<tr>
<td>2</td>
<td>Arsenic</td>
<td>Not Detected</td>
<td>5ppm</td>
</tr>
<tr>
<td>3</td>
<td>Cadmium</td>
<td>Not Detected</td>
<td>0.3ppm</td>
</tr>
<tr>
<td>4</td>
<td>Mercury</td>
<td>Not Detected</td>
<td>1ppm</td>
</tr>
</tbody>
</table>

### Table-6: Total bacterial count

<table>
<thead>
<tr>
<th>S.No.</th>
<th>Analysis</th>
<th>Desmostachya bipinnata (S1)</th>
<th>Permissible limit</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Total Aerobic Microbial count</td>
<td>8965 cfu/gm</td>
<td>100000 cfu/gm</td>
</tr>
<tr>
<td>2</td>
<td>Total Yeast &amp; Mould count</td>
<td>&lt;10 cfu/gm</td>
<td>1000 cfu/gm</td>
</tr>
</tbody>
</table>

### Table-7: Test for aflatoxins

<table>
<thead>
<tr>
<th>S.No.</th>
<th>Test for Aflatoxins</th>
<th>Desmostachya bipinnata (S1)</th>
<th>Permissible limit</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Aflatoxin B1</td>
<td>Absent</td>
<td>0.5ppm</td>
</tr>
<tr>
<td>2</td>
<td>Aflatoxin B2</td>
<td>Absent</td>
<td>0.1ppm</td>
</tr>
<tr>
<td>3</td>
<td>Aflatoxin G1</td>
<td>Absent</td>
<td>0.5ppm</td>
</tr>
<tr>
<td>4</td>
<td>Aflatoxin G2</td>
<td>Absent</td>
<td>0.1ppm</td>
</tr>
</tbody>
</table>
Table-8: TLC Fingerprinting

|--------------------------------------------|------------------------------------------|-----------------------------------------------------|

\[ R_f \text{ value}\]

\[ 0.15, 0.78, 0.81, 0.87 \]

REFERENCE


